

BIOGENETIC-LIKE SYNTHESIS OF d,1-HIRSUTENE

Yasufumi Ohfuno, Haruhisa Shirahama and Takeshi Matsumoto

Department of Chemistry, Faculty of Science,
Hokkaido University, Sapporo, Japan

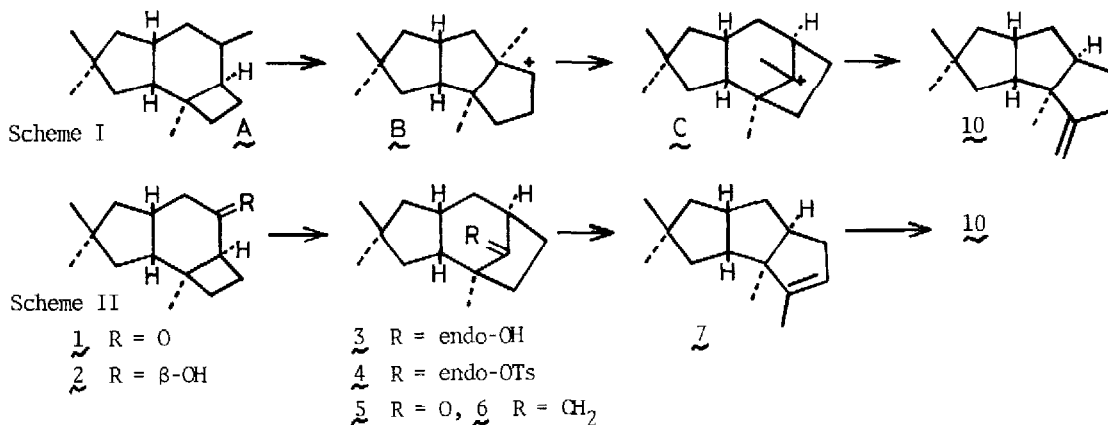
(Received in Japan 22 May 1976; received in UK for publication 21 June 1976)

Hirsutene is a tricyclic sesquiterpene hydrocarbon isolated from *Coriolus consors* and its structure was determined as shown in formula 10¹. The biosynthetic pathway² to hirsutene* has been suggested to involve a sequence of rearrangements depicted in Scheme I. We now wish to describe a chemical synthesis of hirsutene along the proposed biosynthetic route, starting from the previously described 7-keto-13-norprotoilludane 1³.

The keto compound 1 was stereoselectively reduced to 7 β -alcohol 2⁴ with NaBH₄ in EtOH (0°, 2hr, 100%). 2^{5,6}; nmr(CCl₄) δ 0.99, 1.06, 1.16 (each 3H, s), 3.72 (1H, d.d.d., J=4, 7, 11Hz). On treatment with 1 eq. p-TsOH in benzene (80°, 2hr), 2 afforded a mixture of endo-alcohol 3 and its tosylate 4 (40% and 30% respectively). 3^{5,6}; nmr(CCl₄) δ 0.91, 1.01, 1.08 (each 3H, s), 3.52 (1H, d, J=5Hz). 4⁶; (m.p 140°~142°). The tosylate 4 was reduced to 3 with LiAlH₄ in THF (97%, r.t, 15hr). The endo configuration of the hydroxy group of 3 was assigned by the fact that the corresponding ketone 5, obtained by the Jones oxidation of 3 (95%), was reduced quantitatively to the original alcohol 3 with NaBH₄ in EtOH (0°, 2hr)⁷. 5^{5,6}; nmr(CCl₄) δ 0.90, 0.98, 1.10 (each 3H, s); ir(neat) 1750cm⁻¹. Methylenation (r.t, 2hr, ph₃PCH₃Br-tAmONa-benzene) of 5 yielded exomethylene compound 6 (92%), a protohirsutyl cation equivalent. 6^{5,6}; nmr(CCl₄) δ 0.96 (6H, s), 1.06 (3H, s), 4.38, 4.72 (each 1H, broad s). Acid-catalyzed rearrangement of 6 to the hirsutane skeleton was accomplished by treatment with a catalytic amount of p-TsOH in benzene (80°, 20min); an endo isomer 7^{5,6} of hirsutene was produced in 95% yield. 7; nmr(CCl₄) δ 0.96, 0.97, 1.08 (each 3H, s), 1.60 (3H, broad s), 5.01 (1H, broad t, J=2.5Hz); mass 204 (M⁺), 94 (base peak). Epoxydation of 7 (r.t, 30min, m-CPBA-CH₂Cl₂, 98%) afforded endo-epoxide 8^{5,6}, nmr(CCl₄) δ 0.95, 0.98, 1.07, 1.26 (each 3H, s), 3.11 (2H, broad s), which was reduced to tertiary alcohol 9 with LiAlH₄ in THF.

$9^{5,6}$; m.p 49° – 50° ; nmr(CCl_4) δ 0.90, 0.94, 1.05, 1.13 (each 3H, s). Dehydration of 9 ($MsCl$ - Py , 35° , 16hr) gave rise to an isomeric mixture consisting of the starting compound 7 and the desired d,l-hirsutene 10 , in the ratio of 7 to 1, in 70% yield. The mixture was carefully separated by 10% $AgNO_3$ -silica gel chromatography (n-hexane) to give pure 10 ; the hydrocarbon 10 showed completely identical vpc, mass and nmr data with those of natural hirsutene. 10 ; nmr(CCl_4) δ 0.92 (6H, s), 1.05 (3H, s), 4.72 (2H, m); ir(neat) 1655, 878 cm^{-1} ; mass 204 (M^+), 94 (base peak). Since the cis-anti-cis stereochemistry of 1 has been established³, the present synthesis confirms the proposed cis-anti-cis skeleton of hirsutene.

Grateful thanks are given to Dr.S.Nozone, Institute of Applied Microbiology, Tokyo University, for the identification of hirsutene (gc-mass, nmr).



REFERENCES AND FOOTNOTES

- * Hirsutene is suggested to be a biosynthetic precursor of hirsutic acid^{2a,2c} and coriolin^{2b}.
- Structure and synthesis; S.Nozone, J.Furukawa, U.Sankawa and S.Shibata, *Tetrahedron Lett.*, 195 (1976).
 - In the biosynthesis of hirsutene, protoilludyl cation A is assumed as an intermediate. However, no direct evidence has yet been obtained; a) T.C.Feline and G.Mellows, *Chem. Commun.*, 63 (1974); b) M.Tanabe, K.T.Suzuki and W.C.Jankowski, *Tetrahedron Lett.*, 2271 (1974); c) Total synthesis of hirsutic acid; H.Hashimoto, K.Tsuzuki, F.Sakan, H.Shirahama and T.Matsumoto, *Tetrahedron Lett.*, 3745 (1974).
 - Y.Ohfune, H.Shirahama and T.Matsumoto, *Tetrahedron Lett.*, 4377 (1975).
 - A.C.Cope, S.Moon, C.Ho.Park and G.L.Woo, *J.Amer.Chem.Soc.*, 84, 4865 (1962).
 - Obtained as an oil.
 - Satisfactory nmr, ir and mass spectral data as well as elementary analytical values were obtained.
 - A.C.Cope, J.M.Grisar and P.E.Peterson, *J.Amer.Chem.Soc.*, 82, 4299 (1960).